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 lanthanides 10. 2 11.
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 elements 13. 32 14.
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 ionization energy 18. 3
 s² 3p⁴ 19. atomic
 radius 20 ...

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Modern Chemistry 35
 Chapter Test Name
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 Test B, continued Write
 the orbital notation for
 the following elements
 in the space provided
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 chapter 7 test answer
 key. 37. lithium, atomic
 number 3 38. carbon,
 atomic number 6 39
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 Write the ...

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Chapter Test Name
Class Date Chapter
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this figure to answer
questions 7 and 8.

____ 7. A solution
containing 35 g of Li
2SO 4 dissolved in 100
g of water is heated
from 10°C to 90°C.
According to
information in the
figure, this
temperature change
would result in a. an

additional 5 g of Li 2SO
4 in ...

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test b answers. The
measure of the ability
of an atom in a
chemical compound to
attract electrons from
another atom in the
compound is called
_____.

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test b answers. The
measure of the ability
of an atom in a
chemical compound to

attract electrons from another atom in the compound is called _____. 17. The energy required to remove one electron from an atom is called its _____. 18.

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electromagnetic
spectrum.

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in a water molecule is
caused by an uneven
distribution of
electrons between the
oxygen and hydrogen
atoms. 9. The
concentration of H ions
determines whether a
solution is acidic or
basic.

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strong acid is titrated
with a weak base, the
pH of the solution at
the equivalence point
is than 7. 19. A is a
highly purified solid
used to check the
concentration of a
standard solution.
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