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Answer Key Question
Answer 1 B 2 C 3 D 4 C
5 A 6 A 7 C 8 A 9 A 10
C 11 B 12 B 13 D 14 B
15 C 16 B 17 B 18 D 19

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Part B Question 21 (a)
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the use of acids as a
food preservative
related to properties of
acids 2 • Provides a
reason for the use of
acids 1 Sample answer:
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that cause spoilage of
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2013 HSC Chemistry Marking Guidelines Section I, Part B Question 21 (a) Criteria Marks • Provides a reason for the use of acids as a food preservative related to properties of acids 2 • Provides a reason for the use of acids 1

Sample answer: Many microorganisms that cause spoilage of food cannot survive in low pH conditions. The

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Years Early Stage 1
(Kindergarten)

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Working time – 3 hours
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$\text{CH}_3\text{COOH (aq)} + \text{H}_2\text{O (l)} \rightleftharpoons \text{CH}_3\text{COO}^-(\text{aq}) + \text{H}_3\text{O}^+(\text{aq})$ When a small amount of $0.1 \text{ mol L}^{-1} \text{ HCl}$ is added to the buffer solution, $[\text{H}_3\text{O}^+]$ increases, disturbing the equilibrium. According to Le Chatelier's principle, equilibrium shifts to the left to use up H_3O^+ and minimise the disturbance.

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1. Sample answer: In step 1, chloroethane (A) can be converted to ethanol (B) by heating with dilute aqueous potassium hydroxide. The mass spectrum of ethanol will have a single molecular ion peak at $m/z = 46$, whereas chloroethane will give two peaks at m/z 64 and 66.

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guidelines section i	'Maharashtra HSC
part a multiple choice	Board Exam Class XII
answer key question	Solved Question
answer 1 b 2 c 3 d 4 c	