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Chapter 9 Stoichiometry Answers Section

CHAPTER 9 REVIEW Stoichiometry MIXED REVIEW SHORT
 ANSWER Answer the following questions in the space provided. 1.
 Given the following equation: $\text{C}_3\text{H}_4(\text{g}) + x\text{O}_2(\text{g}) \rightarrow 3\text{CO}_2(\text{g}) +$
 $2\text{H}_2\text{O}(\text{g})$ 4 a. What is the value of the coefficient x in this
 equation? 40.07 g/mol b. What is the molar mass of C_3H_4 ? 2
 mol O_2 :1 mol H_2O c. What is the mole ratio of O_2 to H

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Answers CHAPTER 9 REVIEW Stoichiometry SECTION 9-3

PROBLEMS Write the answer on the line to the left. Show all your
 work in the space provided. 1. 88% If the actual yield of a
 reaction is 22 g and the theoretical yield is 25 g, calculate the
 percent yield. 2. 6.0 mol of N_2 are mixed with

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