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Please answer the
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about the combustion
of acetylene: $2\text{C}_2\text{H}_2 + 5\text{O}_2 \rightarrow 4\text{CO}_2 + 2\text{H}_2\text{O}$ 1.
How many moles of
acetylene (C_2H_2) are
needed to completely
react with 12 moles of
oxygen? 2. How many
grams of carbon
dioxide are produced
from the combustion of
1.3 moles of acetylene

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