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Example 12.1.1. The solution in Figure 12.1.1 contains 10.0 g of cobalt(II) chloride dihydrate, $\text{CoCl}_2 \cdot 2\text{H}_2\text{O}$, in enough ethanol to make exactly 500 mL of solution. What is the molar concentration of $\text{CoCl}_2 \cdot 2\text{H}_2\text{O}$? Given: mass of solute and volume of solution Asked for: concentration (M) Strategy: To find the number of moles of $\text{CoCl}_2 \cdot 2\text{H}_2\text{O}$, divide the mass of the compound by its molar ...

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