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Gaseous state of matter: The most mobile state of matter is the gaseous state having the following characteristics: 1. The gas molecules neither have fixed shape nor have fixed volume both being determined by the volume of the container. 2.The gas is highly compressible in nature because of large intermolecular space between them.

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The equation of state for an ideal gas: Those gas which obeys ideal gas equation i.e  $PV = nRT$  is known as ideal gas. At high temperature and low pressure a gas behaves as an ideal gas. Boyle's law states that for a fixed amount of an ideal gas kept at a fixed temperature, pressure and volume are inversely proportional i.e

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The Gaseous State. This is the simplest state of matter. Characteristics of this 'state of matter are. In gases, the intermolecular forces are weakest. Gases are highly compressible. Gases exert pressure equally in all directions. Gases have much lower density than the solids and liquids. The volume and the shape of gases are not fixed.

### CBSE Class 11 Chemistry Notes : States of Matter | AglaSem ...

2. Gaseous exchange relies on diffusion. To be efficient, the gaseous exchange surface must: - thin - shorter distance to diffuse - moist - allow gases to dissolve - large surface area - have a concentration gradient across surface - maintained by movement of air and transport/ use of gas. These features are present in gills (fish) and alveoli (lungs).

### Gaseous exchange - Biology Notes for IGCSE 2014

Gas Exchange. In order to see how gas exchange occurs we need to look further into the lungs. Firstly we know that air enters the body through the mouth or nose, from here it moves to the pharynx (throat), passes through the larynx (voice box) and enters the trachea. The trachea splits into two branches, the left and right bronchus, each bronchus divides many times into smaller branches called ...

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