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Q. The equation for the reaction between magnesium and dilute sulfuric acid is shown.

$$\text{Mg} + \text{H}_2\text{SO}_4 \rightarrow \text{MgSO}_4 + \text{H}_2$$
 The mass of MgSO_4 is 120. Which mass of magnesium sulfate will be formed if 12 g of magnesium are reacted with sulfuric acid?

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