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alpha particle. beta particle. gamma particle. 1) I give the simple definition at the top, usually followed b.... a helium nucleus that is emitted from a radioactive atom. Alph.... an electron that is emitted from a radioactive atom. Beta part.... a high-energy photon (particle of light) that is emitted from....

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answer choices. A plutonium atom is used to start a chain reaction that detonates a nuclear weapon.

A uranium atom is split apart into lighter elements. Two hydrogen atoms are combined to form a helium atom. Two hydrogen atoms bond with an oxygen atom to form a water molecule. Tags:

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Play this game to review Nuclear Chemistry. Two or more nuclei combine to form one larger nucleus in the process of nuclear ____.

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Nuclear chemistry - Chemistry test: 1) Each a- particle has the mass: a) Equal to that of hydrogen. b) Nearly four times that of hydrogen atom. c) Half of the hydrogen atom. d) None of these. ANSWER : Nearly four times that of hydrogen atom. 2) Which among the following has the highest penetrating power? a) γ - rays b) β - rays

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Nuclear Chemistry Review. ... Identify the missing particle in the following nuclear reaction: 99 43 Tc → ____ + 0-1 e ? 99 44 Ru ? 99 43 Tc ? 99 42 Mo ? 99 43 Es; What is the effect of this decay? 99 43 Tc → 99 44 Ru + 0-1 e ? a neutron is converted to a proton ?

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Identify the following based on the mass number and atomic number: 0. 0 A) alpha particle B) beta particle C) positron particle D) gamma ray E) neutron. 7. Gamma rays require the heaviest shielding of all the common types of nuclear radiation because gamma rays have the A) highest energy. B) most intense color.

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Solution: n+p=Mass Number. 90+n=231. n=141. where n is number of neutrons and p is number of protons.Thus ratio between n and p is; n 0 /p + =141/90=1,56. Since ratio is greater than 1, 90231 Th has unstable nucleus and it is radioactive element. 2. A, B, C and D elements form compounds AC, A 2 D and BD.

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CHEM-2001 FINAL EXAM + ANSWERS CHAPTER 18 - NUCLEAR CHEMISTRY RADIOACTIVE DECAY KINETICS Which of the following statements concerning nuclear reactions is false All statements are true except “E”. Note that temperature is not a variable in any of the kinetics equations for nuclear decay (A = λN; ln(A t /A o) = -λt; t 1/2 = 0.693/λ)? any of the kinetics

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