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STRUCTURE OF HYDROCARBONS Draw the structure of the compounds below. J Chemistry IF8766 97 ©Instructional Fair, Inc, 1. ethane 5. t~L -C 5 - 4c*6Lm-r-+h, (jl h,eP vie, -C C C C C C.- 0- 2.~ al W'~- ' I il~~ s'-C- -C s - ff 6. 3, 3-dimethyl pentane Y~ I _ v(3 Li , -cl i e~ln~l - 3, 3 -d i rn~ 4ll~{ Oc.+cLn e v 1. I ~ G IC — I i 1 -C -C - C --C --C 1 ~G- 1 7, 2, 3-dimethyl pentane C c l ~ C ...

NAMING HYDROCARBONS Name - Ms Galloway

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Predict the molecular formula of decane and explain your answer. The formula will be C 10 H 22.This is because n = 10. So, 2n+2 = (2 × 10) + 2 = 20 + 2 = 22.

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Many hydrocarbons occur in nature. In addition to making up fossil fuels, they are present in trees and plants, as, for example, in the form of pigments called carotenes that occur in carrots and green leaves. More than 98 percent of natural crude rubber is a hydrocarbon polymer, a chainlike molecule consisting of many units linked together. The structures and chemistry of individual ...

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One way around this is to use a condensed structure, which lists the formula of each C atom in the backbone of the molecule. For example, the condensed structure for ethane is CH_3CH_3 , while it is $\text{CH}_3\text{CH}_2\text{CH}_3$ for propane. Table 16.1 “The First 10 Alkanes” gives the molecular formulas, the condensed structural formulas, and the names of the first 10 alkanes. Table 16.1 The First 10 ...

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Chemistry IF8766 96 ©instructional Fair, Inc. Name m, e, p, b STRUCTURE OF HYDROCARBONS Draw the structure of the compounds below. J Chemistry IF8766 97 ©Instructional Fair, Inc, 1. ethane 5. $\text{t}\sim\text{L}-\text{C}_5-4\text{c}^*\text{6Lm-r-+h}$, (jl h,eP vie, -C C C C C C.- 0- 2.~ al W'~- ' I il~~ s'-C- -C s - ff 6. 3, 3-dimethyl pentane ... NAMING HYDROCARBONS Name - Ms Galloway Tomorrow's answer's today! Find correct ...

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