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equation. a way to
describe a chemical
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Chapter 8. A number in
front of a chemical
formula in an equation
that indicates how
many molecules or
atoms of each reactant
and product are
involved in a reaction.
When the positive ion
of one compound
replaces the positive
ion of another to form
two new compounds.

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Chapter 8. STUDY. PLAY. activity series. a list of elements organized according to the ease with which the elements undergo certain chemical reactions. chemical equation. represents, with symbols and formulas, the identities and relative amounts of the reactants and products in a chemical reaction.

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CHAPTER 8 REVIEW

Class SĖc7îB-n

Chemical Equations
and Reactions SHORT

ANSWER Answer the following questions in the space provided. 1. Match the symbol on the left with its appropriate description on the right. (aq) (a) (b) (c) (d) (f) A precipitate forms. A

gas forms. A reversible reaction occurs. Heat is applied to the reactants.

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Chemical Equations
and Reactions SECTION
2 SHORT ANSWER

Answer the following questions in the space provided. 1. Match the equation type on the left to its representation on the right. c synthesis (a) $AX + BY \rightarrow AY + BX$ d decomposition (b) $A + BX \rightarrow AX + B$ b single-displacement (c) $A + B \rightarrow AX$ a double-displacement (d) $AX \rightarrow A + X$

8 Chemical Equations and Reactions

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