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CHAPTER 1. INTRODUCTION TO RESERVOIR ENGINEERING. PROBLEM 1.1 Calculate the volume 1 lb-mole of ideal gas will occupy at: a) 14.7 psia and 60°F b) 14.1 psia and 32°F c) 14.7 plus 10 oz and 80°F ...

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Solution: Writing down the kinetic and potential energy yields: $T = \frac{1}{2} m l^2 \dot{\theta}^2$, $U = \frac{1}{2} k x^2 + mgh$. $U = \frac{1}{2} k l^2 \sin^2 \theta + mgl(1 - \cos \theta)$ Here the spring deflects

a distance $l \sin \theta$, and the mass drops a distance $l(1 - \cos \theta)$. Adding up the total energy and taking its time derivative yields: $\frac{d}{dt} \left(\frac{1}{2} m l^2 \dot{\theta}^2 + \frac{1}{2} k l^2 \sin^2 \theta + mgl(1 - \cos \theta) \right) = 0$

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