
Sleeping With The Enemy Book

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This agrees with Schroeder's Figure 1.14. To put in some numbers, we can use the data from the earlier problem. For water, the values given by Schroeder are (at $25^\circ\text{C} = 298\text{ K}$): $\rho = 2.57 \times 10^4\text{ kg m}^{-3}$ (20) $T = 4.52 \times 10^{10}\text{ Pa}$ (21) For one mole of water, the volume is $18.068 \times 10^{-6}\text{ m}^3$, so $C_P C_V = (298) 18.068 \times 10^{-6} 2.57 \times 10^4 2 4.52 \times 10^{10} (22) = 0.787\text{ J K}^{-1} (23) \text{ C}$

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