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have on the environment and the species living within it. The four major pillars of the environment are the biosphere, the atmosphere, the hydrosphere, and the geosphere.

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Thermodynamics (Eq. (1)) states that the change in internal energy ( $\Delta E$ ) of a system is the sum of heat flow into the system ( $q$ ) and work done on the system ( $w$ ), thus describing the  $\Delta E$  that a given system experiences:  $\Delta E = q + w$

$w(1)$

### **Fundamental Environmental Chemistry - EOLSS**

42 Chapter 2 Environmental Chemistry Section 2.2 Stoichiometry 43 We already know that there are 44 grams per mole of  $\text{CO}_2$  so we do not need to recalculate that. Two moles of butane ( $2 \text{ mol} \times 58 \text{ g/mol} = 116 \text{ g}$ ) yields 8 moles of carbon dioxide ( $8 \text{ mol} \times 44 \text{ g/mol} = 352 \text{ g CO}_2$ ), So we can set up the following proportion:  $116 \text{ g C}_4\text{H}_{10} : 100 \text{ g C}_4\text{H}_{10}$

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