

Dr Khalil Yaich Avis

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Bond Energy Worksheet Answers 1. ΔH= Bonds Broken – Bonds Formed = [H-H + Cl-Cl]-[2(H-Cl)] = [436 kJ/mol+242 kJ/mol] – [2(431)] = -184 kJ/mol 2. ΔH= Bonds Broken – Bonds Formed = [4 H-C + C=C + F-F] – [C-C + 4 C-H + 2 C-F] = [4(413) + 614 + 155]-[348 + 4(413) + 2(485)] = 2421 – 2970 kJ/mol = -549 kJ/mol 3.

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Bond Energies - Chemistry LibreTexts The bond energies are positive. B. The bonds are bond breaking. 8) a. The units for the bond energy is kJ/ mol B. 1 mole bonds. 9) No, there are isnt a double bond because 1. C-C 340 C=C 602 2. C-O 358 C=O 745 10) There is the same absoulte value, but different sign. ... Download Pogil Bond Energy Answer Key ...

Bond Energy Pogil

Energy in = 2 × 366 = 732 kJ/mol. Energy out = 436 + 193 = 629 kJ/mol. Energy change = in - out = 732 – 629 = +103 kJ/mol. The energy change is positive, showing that energy is taken in from ...

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The energies you calculated in Questions 5c and 6a above are called bond energies. The bond energy for a particular type of bond can vary from one molecule to another because the atomic environment of a bond can influence the amount of energy needed to break the bond. For example the carbon-carbon

Bond Energy - Mrs. allen's chemistry class

a. When bonds are broken there is a positive energy change. b. Breaking bonds (endothermic). c. When bonds are formed there is a negative energy change. d. Forming bonds is (exothermic). 4. Find two reactions in Model 1 that are exact opposites of each other, that is, one reaction is the reverse of the other reaction. a.

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A good answer should state that as the distance between a proton (nucleus) and electron increases, the attractive force decreases. This is followed by a quick class discussion to summarize the previous day’s lesson and leads quickly into model 3 and 4 of the POGIL activity.

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