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mass of the reactants equals the mass of the products. When a solid produced by a chemical reaction separates from the solution it is called... a precipitate.

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STUDY. PLAY. Kinetic Molecular Theory.

1) gas molecules are small, hard spheres with an insignificant volume 2) the motion of the particles in a gas is random, constant, and rapid 3) all collisions between particles in a gas are perfectly elastic - they are not attracted or repelled by each other

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of most gases is terms of expansion,

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o Know the difference between directly and inversely proportional.
o Know that the pressure and volume of a gas are inversely proportional.
o Know how to use Boyle's Law to solve for the initial or final pressure or volume of a.

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questionMole answerThe amount of a substance whose mass is equal to the atomic mass in grams
questionAvogadro's number.
answer 6.022×10^{23}

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