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Question 6. Define the following terms: (a) Atomic number (b) Mass number (c) Nucleons (d) Valence shell Answer: (a) Atomic number: Atomic number refers to the number of protons present in an atom. It is denoted by Z. Example: An atom of oxygen contains 8 proton Therefore its atomic number is 8.

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Simplified Chemistry Physics Chemistry Biology Maths Geography
HistoryCivics. Exercise. Question 1.(1984) Define the terms :
atomic number, mass number and electron ...

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3 Atomic structure 1. An atom is the smallest particle of an
element that can exist by itself and still have the same chemical
properties as the element. 2. - Protons - Neutrons - Electrons
Protons and neutrons have a relative mass of 1, whereas
electrons have a relative mass of 1/1840. Protons have a relative
charge of +1, neutrons have no

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Question Answer Explanation 1 C Gas syringe is for measuring
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