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Chemistry/Chapter#16/Hydrocarbons/Lecture#02(2/2)/Nomenclature of Alkanes

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16)HZ is a weak acid. An aqueous solution of HZ is prepared by dissolving 0.020 mol of HZ in sufficient water to yield 1.0 L of solution. The pH of the solution was 4.93 at 25.0 °C. The K_a of HZ is _____. A) 9.9×10^{-2} ... Chapter 16. Practice Questions.tst

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Chapter 16. Practice Questions

Practice-Problem-6660020 book December 22, 2010 8:37 Chapter 16: Colligative Properties of Solutions 45 16-4. The mole fraction of $(\text{NH}_4)_2\text{SO}_4(\text{aq})$ is given by $x_{(\text{NH}_4)_2\text{SO}_4} = \frac{n_{(\text{NH}_4)_2\text{SO}_4}}{n_{(\text{NH}_4)_2\text{SO}_4} + n_{\text{H}_2\text{O}}}$ Because $(\text{NH}_4)_2\text{SO}_4(\text{aq})$ is a strong electrolyte, it dissociates completely into $\text{NH}_4^+(\text{aq})$ and $\text{SO}_4^{2-}(\text{aq})$ ions. Assume a one ...

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statement is true and if the statement is false. 1) According to the Arrhenius theory, a neutralization reaction involves the combination of hydrogen ions and hydroxide ions to form water.

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