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Surface tension and surface thermodynamic properties of

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Thus the energy consumed is at least the thermodynamic surface energy W_s arising from the specific thermodynamic surface energy Γ , $W_s = 4\pi r_0^2 \Gamma$. At the critical equilibrium point, the closing stress $2\Gamma/r_0$ due to the surface tension acting at the boundary of the cavity is equal to the hydrostatic tension ρh , and hence $r_0 = 2\Gamma/\rho h$.

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length, or of energy per unit area. The two are equivalent, but when referring to energy per unit of area, it is common to use the term surface energy, which is a more general term in the sense that it applies also to solids.

Surface tension - Wikipedia

studies of surface-tension, $G_s = \gamma$. Temperature dependence can be determined by differentiating with respect to temperature. • Thus, temperature dependence of surface tension leads to the specific surface entropy. • Temperature dependence of γ predicting by a semi-empirical equation (van der Waals and Guggenheim: where $\gamma_0 = \gamma$ at $T = 0$ K, T

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From 251 data points examined the average relative deviation (ARD) of the correlated and calculated densities and surface tension from the experimental ones was found to be 1.63% and 2.46%. Besides, some surface thermodynamic functions such as the surface entropy (S_s) and surface enthalpy (H_s) of studied liquids were also computed via our method. The ARD (in %) were found to be equal to 3.94 and 2.44, respectively.