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Sample Exam 1 Name Chem

1 Chem. 111 Name \_\_\_\_ Spring 07 EXAM 1 (Last) (First) 100 Points All significant figures and set-ups must be shown in an organized way. All units must be shown throughout the problems. Use the following table for molar masses. Compound Molar Mass g/mole Compound Molar Mass g/mole H 2 SO 4 98.0795 K 3PO

Chem. 111 Name EXAM 1 (Last) (First) - Cerritos College

Exam 1 Name \_\_\_\_ CHEM 212 1. (8 pts) (8 pts) For each of the following compounds, indicate the number of resonances (not multiplicities) that would be observed on a 1H NMR spectrum: a.

Exam 1 Name CHEM 212 - Penn State Behrend

Exam 1 Name \_\_\_\_ CHEM 210 1. (6 pts) Give each non-hydrogen atom the proper formal charge, even if zero; all electrons are shown. a. b. c. d. 2. (9 pts) Draw one valid Lewis structure for each compound. Assign formal charges to atoms if necessary. a. nitromethane, CH 3 NO 2 b. ozone, O 3 c. hydrazine, N 2 H 4 3.

Exam 1 Name CHEM 210 - Pennsylvania State University

UIN# \_\_\_\_ Name\_\_\_\_ Chemistry 119 Practice Exam 1 Please make sure that you are seated in the correct seat. Write your name and UIN in the spaces above. Read the directions and the honor code statement before signing your name on the signature line below. There are 18 multiple choice questions on this exam.

Practice Exam 1.docx - UIN Name Chemistry 119 Practice ...

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Science Quiz: Chemistry: Naming Compounds

Name CHEM 2425 Practice Exam 1 Chapters 12—16 Name the following compounds: Chl z . c S03 (excess CHCUZCU C-c( 8. Complete the following reactions: c 1-13 cÃ3 4 c 143 c l b-cv13 A (C 113 ucl The following reactions are incorrect as written. Briefly indicate what is wrong with each.

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#### Naming ionic compounds (practice) | Khan Academy

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#### Chemistry Olympiad Exams - American Chemical Society

Chemistry Self-Assessment & Reflection: Exam #1 Name: \_\_\_\_ DUE: At the next class meeting, hand in this completed form at the beginning of lecture. This form will help you to analyze your exam performance and find strategies that work best for you in learning the material for this course.

#### Chemistry Self-Assessment & Reflection: Exam #1 Name: DUE ...

For example, hydrogen contains a single proton ( $Z = 1$ ), whereas helium contains two ( $Z = 2$ ). Atoms of a single element may differ in terms of the number of neutrons in their atomic nuclei, however. The total number of protons and neutrons in an atom is referred to as the atomic mass, or  $M$ . Helium typically has an atomic mass equal to 4, but there is another helium isotope for which  $M = 3$ .

#### Chemistry Practice Questions - Study Guide Zone

1 Chem 1C Midterm 1 Practice Test 1 Name \_\_\_\_ Perm Number \_\_\_\_ All work must be shown on the exam for partial credit. Points will be taken off for incorrect or missing units. Calculators are allowed. Cell phones may not be used for calculators. On fundamental and challenge problems you must show your work in order to receive credit for the ...

#### Chem 1C Midterm 1 of last name Practice Test 1

Examination specifications and sample materials. Adjusted examination specifications for 2020 (pdf - 128.01kb) (June 2020) Sample written examination (pdf - 1.43mb) (Version 2, April 2017) Please note: The sample examination has not been amended to reflect the adjustments made to the study design in 2020.

#### Pages - Chemistry

CHEM 1411 Sample Exam # 1 Name: \_\_\_\_ (chapters 13,14, and 15) Score: Directions- please answer the following multiple-choice questions next to each number. Please show all your work for bonus question and part(2) questions in the space provided.

#### CHEM 1412 Sample Exam 1 - learning.hccs.edu

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