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Solution is a homogeneous mixture of two or more substances in same or different physical phases. The substances forming the solution are called components of the solution. On the basis of number of components a solution of two components is called binary solution. Solute and Solvent In a binary solution, solvent is the component which is present in large quantity while the other

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$2 \text{Fe(s)} \rightarrow 2 \text{Fe}^{2+} + 4\overline{\text{e}}$ $E^\circ = -0.44 \text{ V}$. Electrons released at anodic spot move through metal and go to another spot on the metal and reduce O_2 in presence of H^+ . This spot behaves as cathode. $\text{O}_2(\text{g}) + 4 \text{H}^+(\text{aq}) + 4\overline{\text{e}} \rightarrow 2 \text{H}_2\text{O(l)}$ $E^\circ = 1.23 \text{ V}$. The overall reaction is, $2 \text{Fe(s)} + \text{O}_2(\text{g}) + 4 \text{H}^+(\text{aq}) \rightarrow 2 \text{Fe}^{2+} + 2 \text{H}_2\text{O(l)}$ $E^\circ = 1.67 \text{ V}$

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Students can Download Chapter 13 Amines Notes, Plus Two Chemistry Notes helps you to revise the complete Kerala State Syllabus and score more marks in your examinations. Kerala Plus Two Chemistry Notes Chapter 13 Amines. Amines Derivatives of NH_3 , obtained by replacement of one, two or all the three hydrogen atoms by alkyl and/or aryl groups, e.g. $\text{CH}_3 - \text{NH}_2$, $\text{C}_6\text{H}_5\text{NH}_2$, $\text{CH}_3 - \text{NH} - \text{CH}_3$...

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OH groups are directly attached to carbon atom (s) of an aliphatic system.

Plus Two Chemistry Notes Chapter 11 Alcohols, Phenols and ...

Kerala Plus Two Chemistry Notes Chapter 4 Chemical Kinetics. Chemical kinetics is the branch of chemistry which deals with the study of the velocity of chemical reactions and their mechanism. Rate of a Chemical Reaction : amount of chemical change per unit time. Average Rate of Reaction:

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Kerala Plus Two Chemistry Notes Chapter 5 Surface Chemistry. Surface chemistry deals with phenomena that occur at the surfaces or interfaces. Adsorption: accumulation of molecular species at the surface rather than in the bulk of a solid or liquid, it is a surface phenomenon, e.g. Moisture gets adsorbed on silica gel. Adsorbate:

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