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Electron Configuration Make a Foldable to help you summarize the three rules that define how electrons are arranged in an atom. STEP 1 Fold a sheet of paper in half lengthwise. Make the back edge about 2 cm longer than the front edge. STEP 2 Fold into thirds.

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How many protons, neutrons, and electrons are in the atom discussed in Questions 6 and 7? Protons: Neutrons: Electrons: Atoms of Six Elements Name Symbol Atomic Number Number of Protons Number of Electrons Hydrogen H 1 Helium He 2 Lithium Li 3 Boron B 5 Carbon C 6 Oxygen O 8 2 1 1 2 5 3 6 3 5 6 88 mass number 4 5 neon-22 10 12 10 22 Ne 10 mass ...

5 ATOMIC STRUCTURE AND THE PERIODIC TABLE

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Electrons In Atoms Guided Answers

Atoms. Matter is anything that takes up space and has mass. All matter is made of atoms. Atoms are the basic building blocks of matter. They make up everything around us; Your desk, the board, your body, everything is made of atoms! Atoms are too small to see without powerful microscopes.

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Arrangement of Electrons in Atoms SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. State the Pauli exclusion principle, and use it to explain why electrons in the same orbital must have opposite spin states. The Pauli exclusion principle states that no two electrons in an atom may have the same set of four quantum numbers.