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chance discoveries. (c) is driven by curiosity. (d) is done for the sake of learning something new.

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CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original

design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

5 The Periodic Law

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CHAPTER 13 REVIEW Ions in Aqueous Solutions and Colligative Properties SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. Use the guidelines in Table 1 on page 437 of the text to predict the solubility of the following compounds in water: soluble a. magnesium nitrate insoluble b. barium sulfate insoluble c. calcium ...

13 Ions in Aqueous Solutions and Colligative Properties

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CHAPTER 6 REVIEW Chemical Bonding SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. a A chemical bond between atoms results from the attraction between the valence electrons and of different atoms. (a) nuclei (c) isotopes (b) inner electrons (d) Lewis structures 2. b A covalent bond consists of (a) a shared electron.

6 Chemical Bonding

CHAPTER 9 REVIEW Stoichiometry MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Given the following equation: $C_3H_4(g) + xO_2(g) \rightarrow 3CO_2(g) + 2H_2O(g)$ 4 a. What is the value of the coefficient x in this equation? 40.07 g/mol b. What is the molar mass of C_3H_4 ? 2 mol O 2:1 mol H 2O c. What is the mole ratio ...

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