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Chapter 12 Stoichiometry Test B

Answer Key

Answer: 4.93×10^{-5} L or 49.3 μ L In

Example 12.2.1 and Example 12.2.2, the identity of the limiting reactant has been apparent: $[\text{Au}(\text{CN})_2]^-$, LaCl_3 , ethanol, and para -nitrophenol. When the limiting reactant is not apparent, we can determine which reactant is limiting by comparing the molar amounts of the reactants with their coefficients in the balanced chemical equation, just as ...

Chapter 12.2: Stoichiometry of Reactions in Solution ...

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