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Flame Test Kit Materials Included in Kit Calcium chloride, $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$, 50 g Sodium chloride, NaCl , 50 g Cupric chloride, $\text{CuCl}_2 \cdot 2\text{H}_2\text{O}$, 50 g Strontium chloride, $\text{SrCl}_2 \cdot 6\text{H}_2\text{O}$, 50 g Lithium chloride, LiCl , 50 g Wooden splints, pkg/100 Potassium chloride, KCl , 50 g Additional Materials Needed (for each lab group)

CF#5607 Flame Test Kit SLK - Science Notebook

Chapter 5 Acids, Bases, and Acid-Base Reactions 159 It's test day in chemistry class—they've been learning about acids and bases—and Fran unwisely skips breakfast in order to have time for some last-minute studying.

Acids, Bases and Acid-Base r

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CF#5607 Flame Test Kit SLK of this laboratory is to construct a table listing the reduction potentials of a series of metal ions, in order of ease of reduction. The series of microscale half-cells is constructed by placing a piece of metal into a 1.0 M solution of its ions for each metal in the series.

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(A) Water acts as an acid in both reactions. (B) Water acts as a base in both reactions. (C) Water acts as an acid in reaction X and as a base in reaction Y. (D) Water acts as a base in reaction X and as an acid in reaction Y. ____13. How many milliliters of 0.010 M NaOH are required to exactly neutralize 20.0 milliliters of 0.020 M HCl ?

Practice Test: Acids & Bases

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