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ThermoChemistry (Energy) WebQuest. Name ____ Period ____ Reaction 1: Reaction 2: Reaction 3: $H^\circ = H^\circ = H^\circ =$ Net Reaction: $\text{SnCl}_2 (\text{s}) + \text{TiBr}_2 (\text{s}) + \text{Cl}_2 (\text{g}) \rightarrow \text{SnBr}_2 (\text{s}) + \text{TiCl}_4 (\text{l})$ $H^\circ \text{ Net} =$ Part 4-

Specific Heat. <http://hyperphysics.phy-astr.gsu.edu/hbase/thermo/spht.html#c1>

Thermochemistry Webquest worksheet - Quia

1) Write the formula / equation that can be used to calculate the heat energy being absorbed or released in a system. 2) Explain each part / variable in the equation. $q = c \cdot m \cdot \Delta T$ 3) Heat Capacity Formula Questions. Read and show the set up / solutions for the two practice questions on this page.

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1910's-1920's? silent movies 2. Name 2 film stars of this era.

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"INTRODUCTION TO ENERGY" WORKSHEET

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Friday: Read Chapter 19 complete page 593 (1-8), 593 (9-10), 599 (11-12), 601-602 (13-16) and 604(17-21)

Chemistry - Mrs. Rydzinski's Science Classroom

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